

Review of RF photoinjector for radiation chemistry

Univ. Tokyo

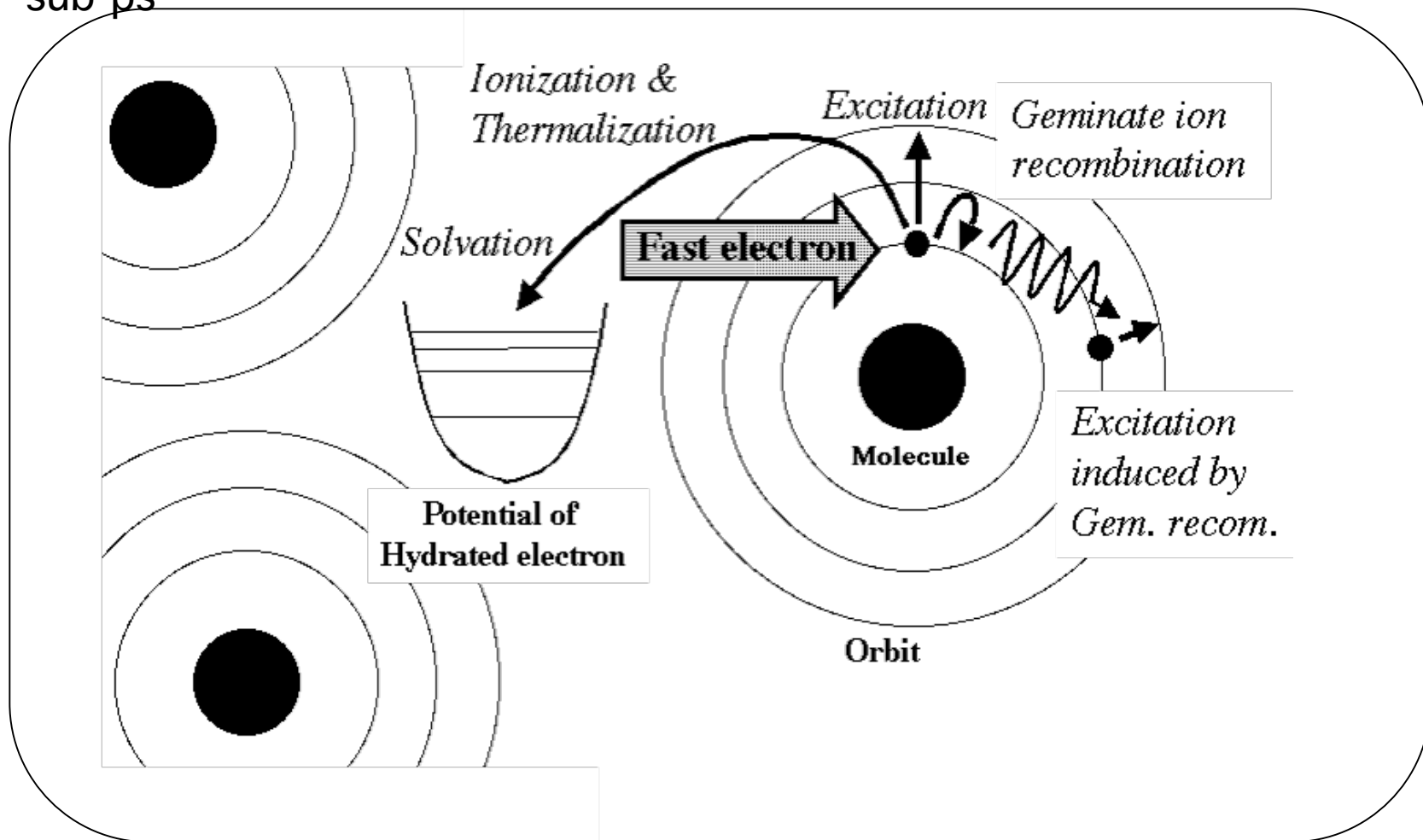
A. Sakumi, M. Uesaka, Y. Muroya, Y. Katsumura

Application for ultra-short pulse

– Radiation Chemistry experiments

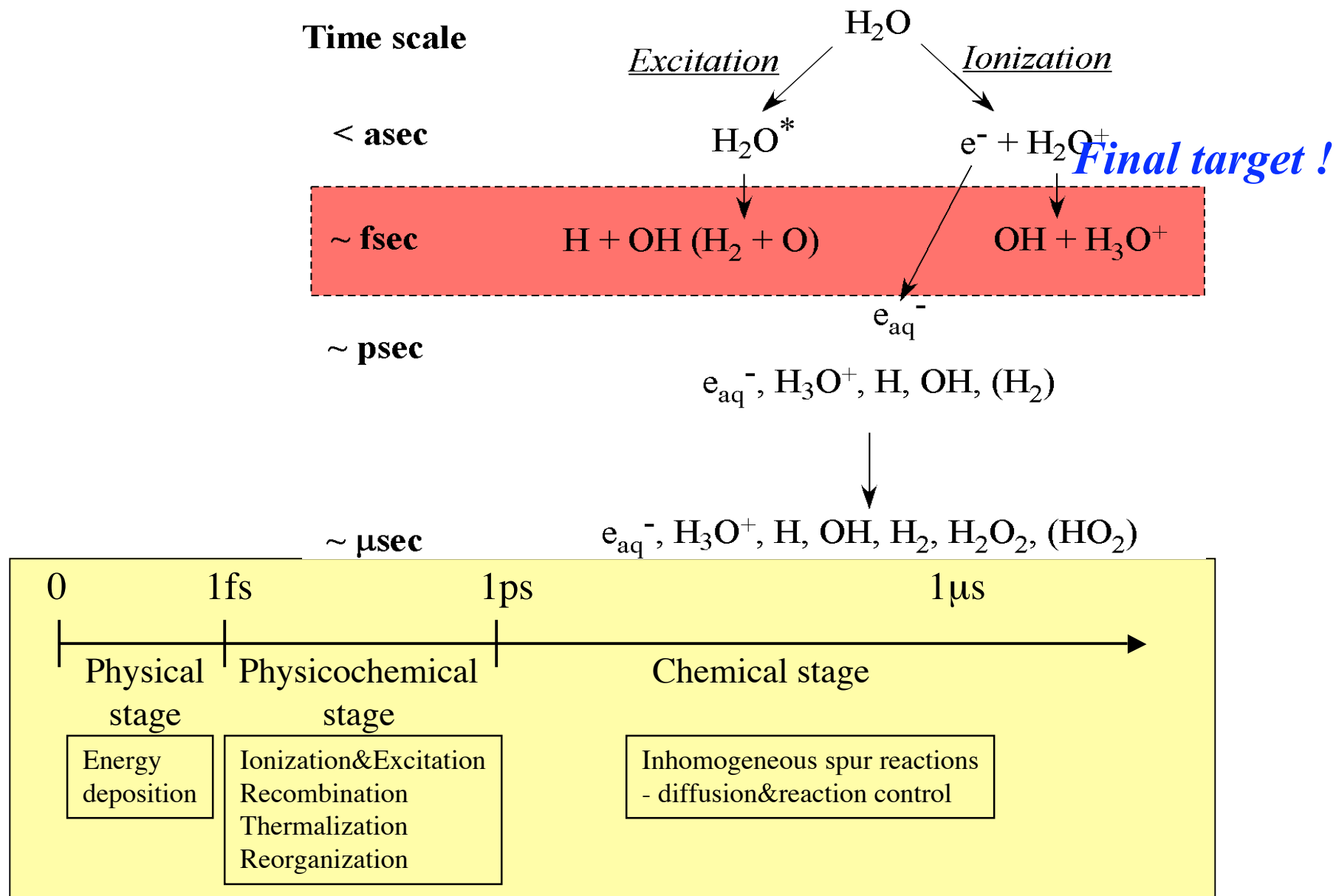
Purpose of the sub-ps pulse radiolysis

- Investigation of the elementary process of radiation induced phenomena which occur in the time scale of ps, even sub-ps



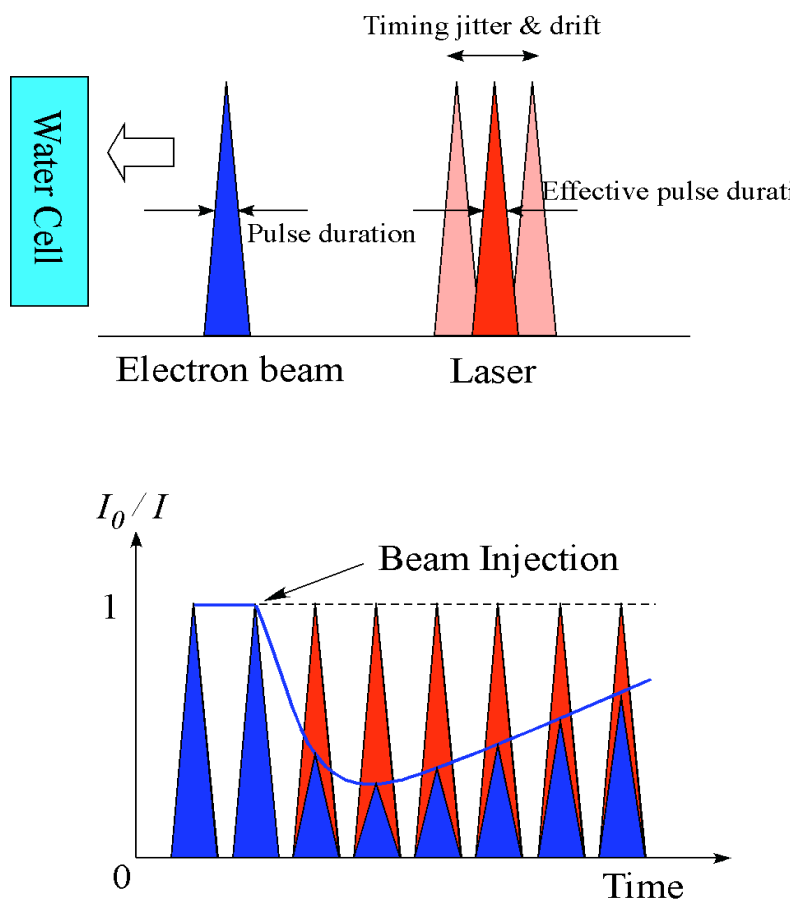
Application for ultra-short pulse

– Radiation Chemistry experiments

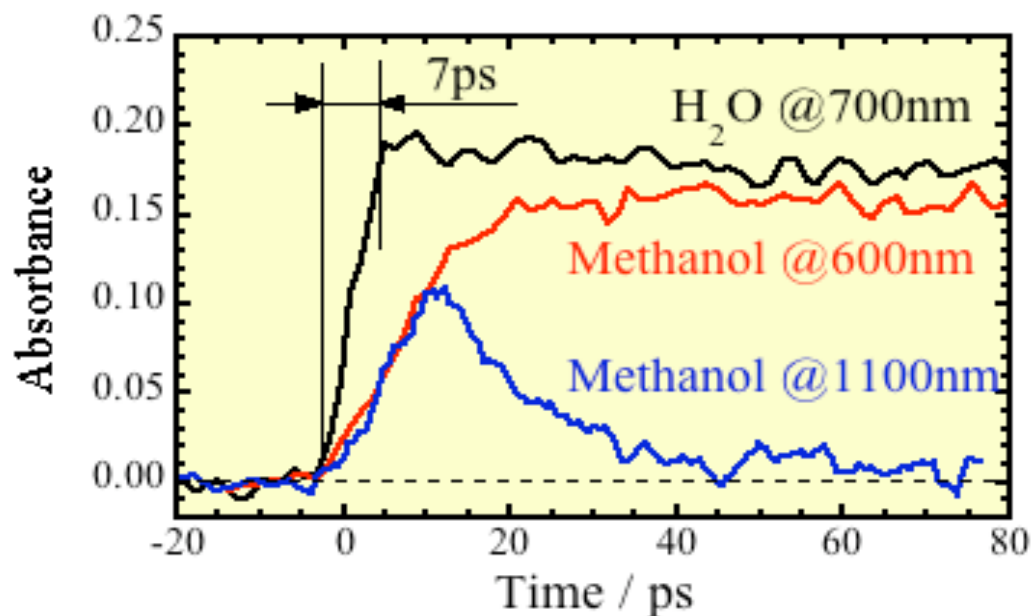


Radiation Chemistry

Pulse radiolysis method



Chemical reaction of water



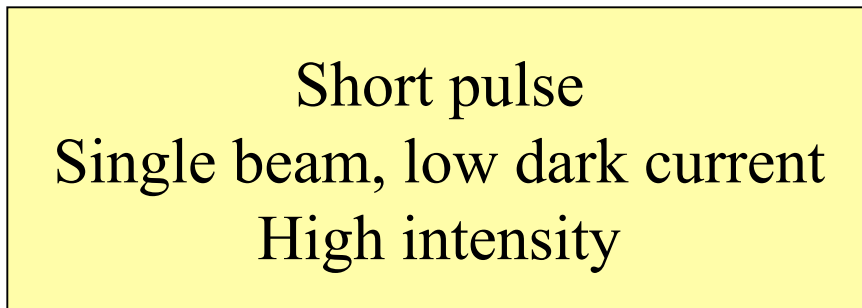
NERS U. Tokyo Y. Muroya et al.,

Requirements

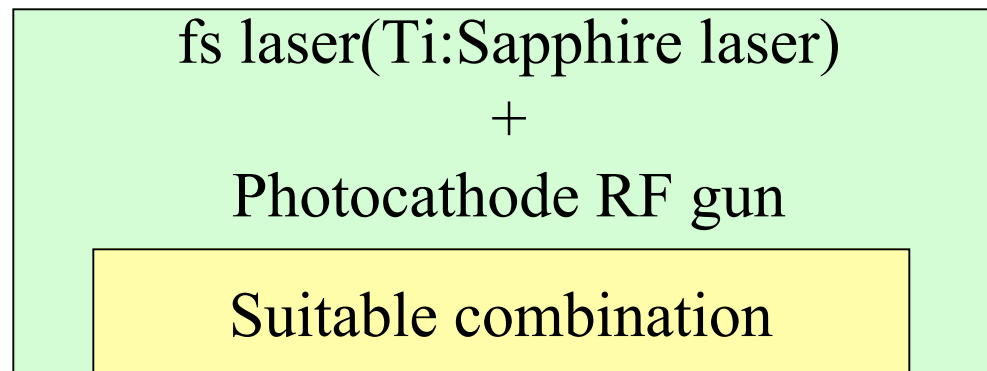
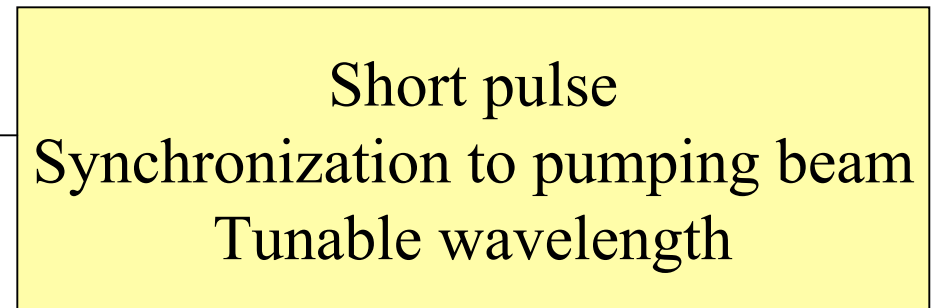
Pulse radiolysis in a time range of sub-picosecond

- I Ultra-short bunch and laser
- II Stable synchronization
- III Intense electron bunch

For Pumping beam



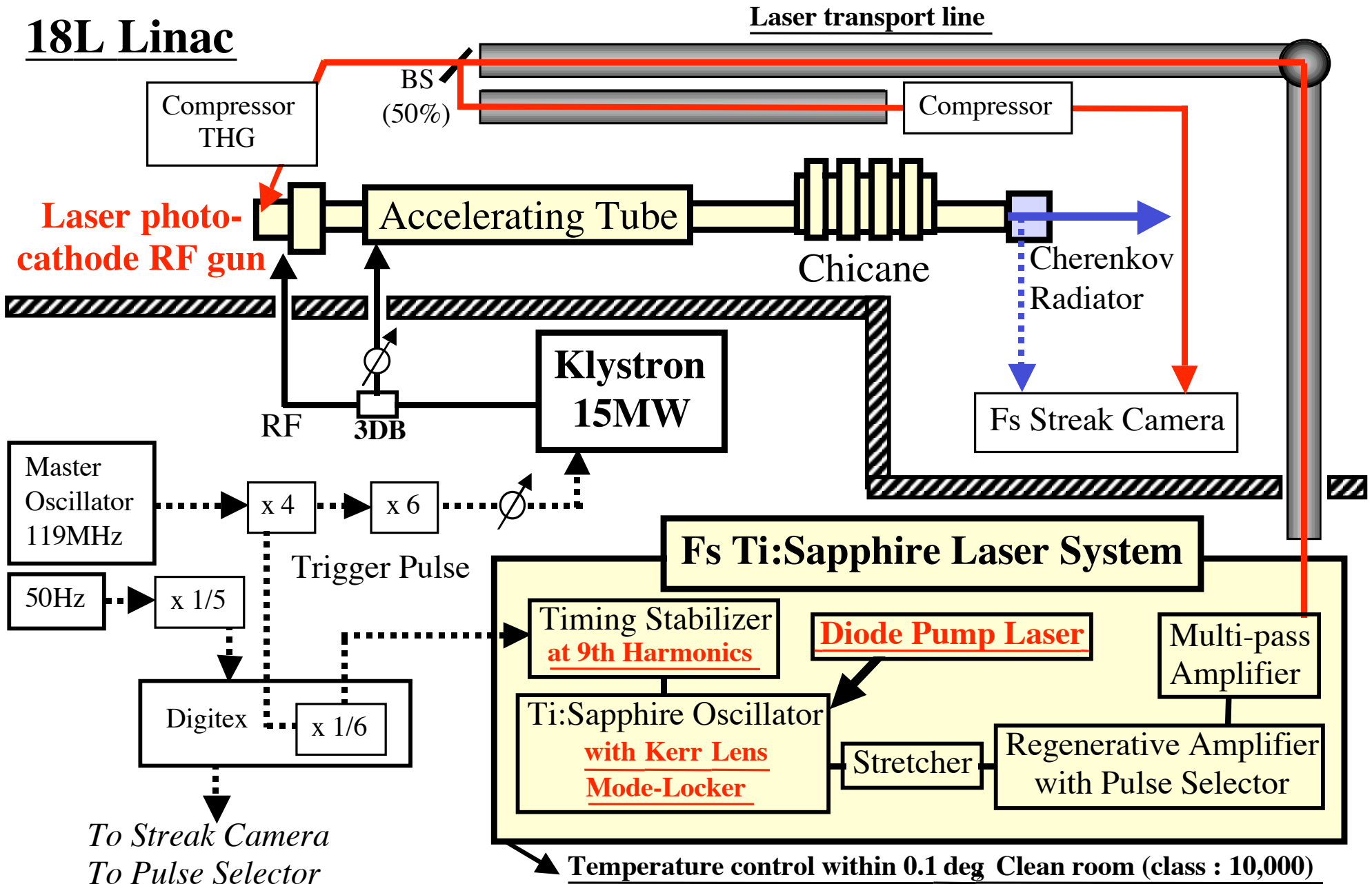
For Probe beam



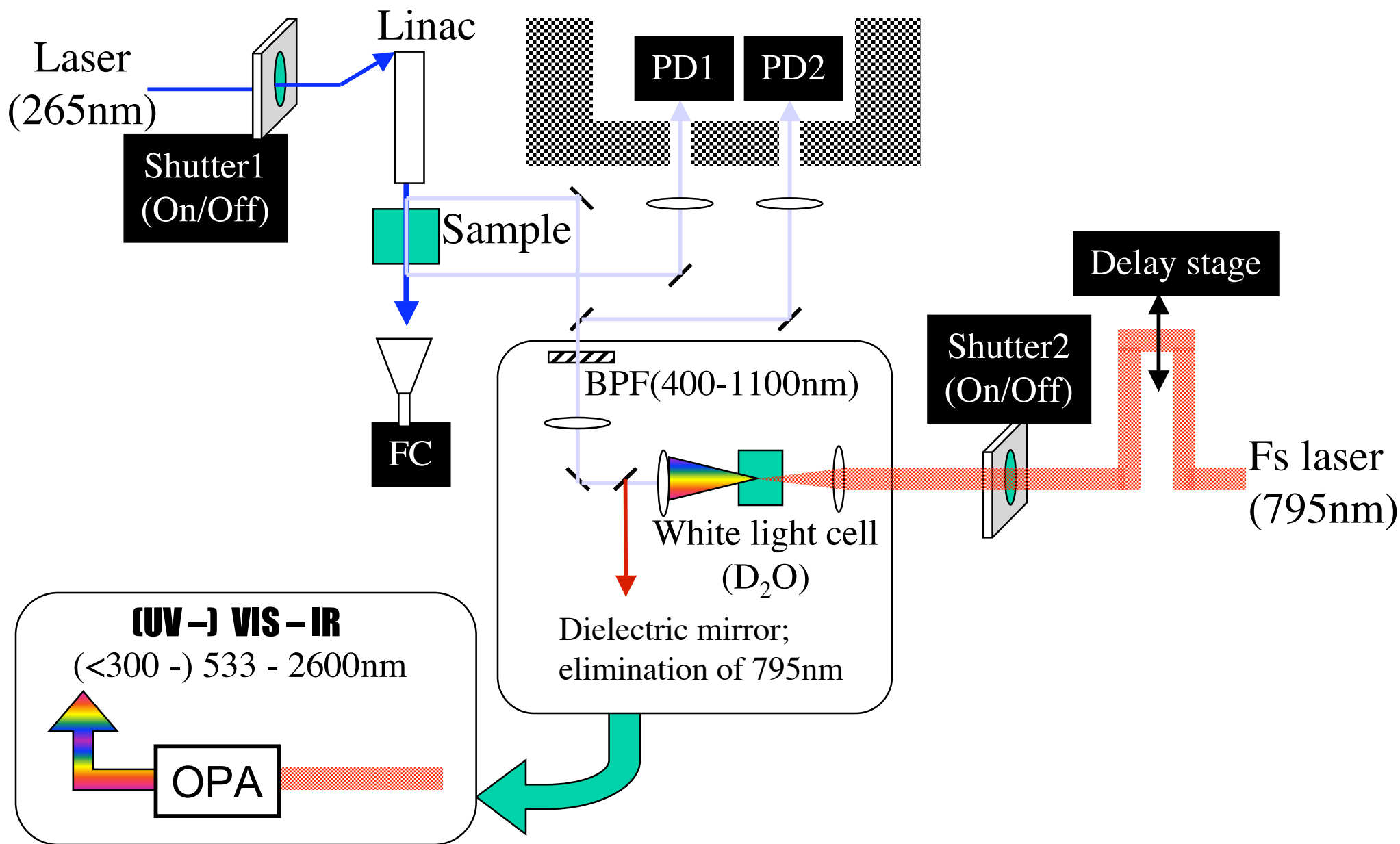
Precise Synchronization System at UTNS

Beam-Material Interactions

www.utns.jp/~beam



Measurement system



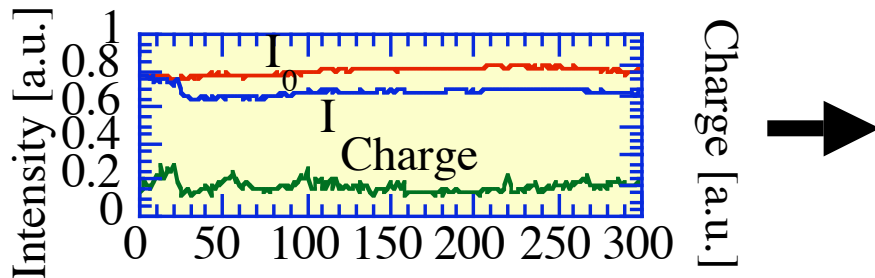
Preliminary Pulse Radiolysis; 795 nm

Condition

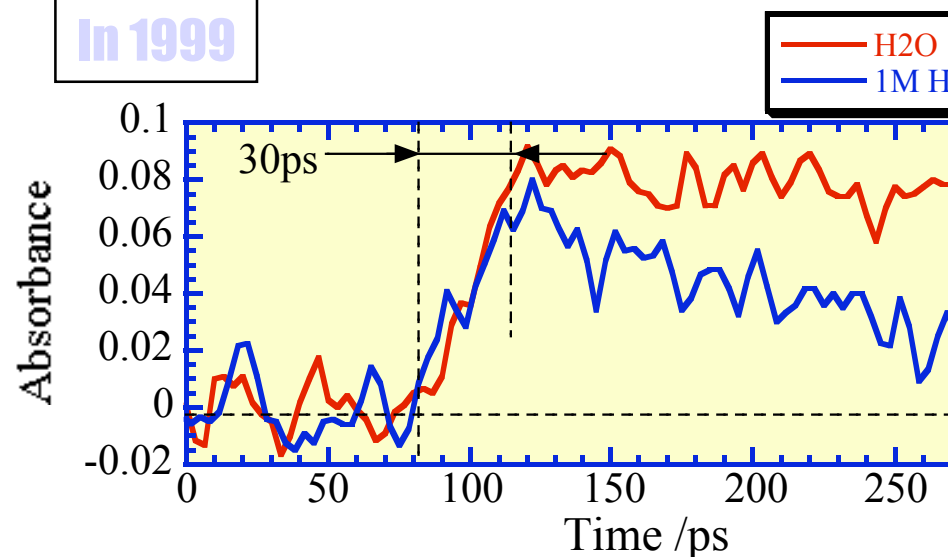
	H ₂ O&1M H ⁺	H ₂ O
// mm	20 mm	5 mm
Charge	0.8-1.0nC	0.8-1.0nC
Beam size	4mm	4mm
Pulse width	7ps	3ps
Wavelength	795nm	
Average	64	16
Time resol.	30ps	9ps

Results

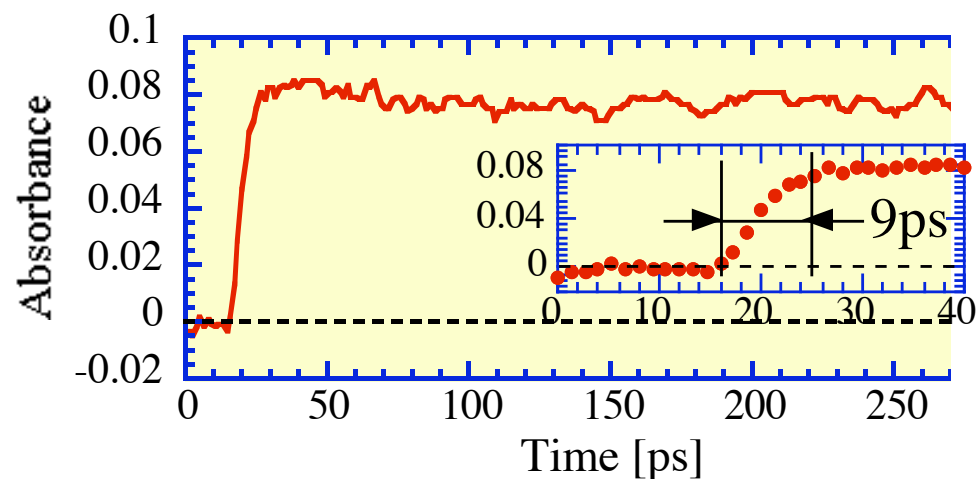
- O.D. still low



In 1999



In 2002



- Time behaviors of e_{aq}^- at 700nm

Results

l/mm	10	5	2	1
O.D.	0.32	0.19	0.08	0.04
S/N	15	10	5	3
Dose	40Gy	47Gy	50Gy	50Gy
Time resol. /ps	12-13ps	6-7ps	4-5ps	<4ps

Good agreement

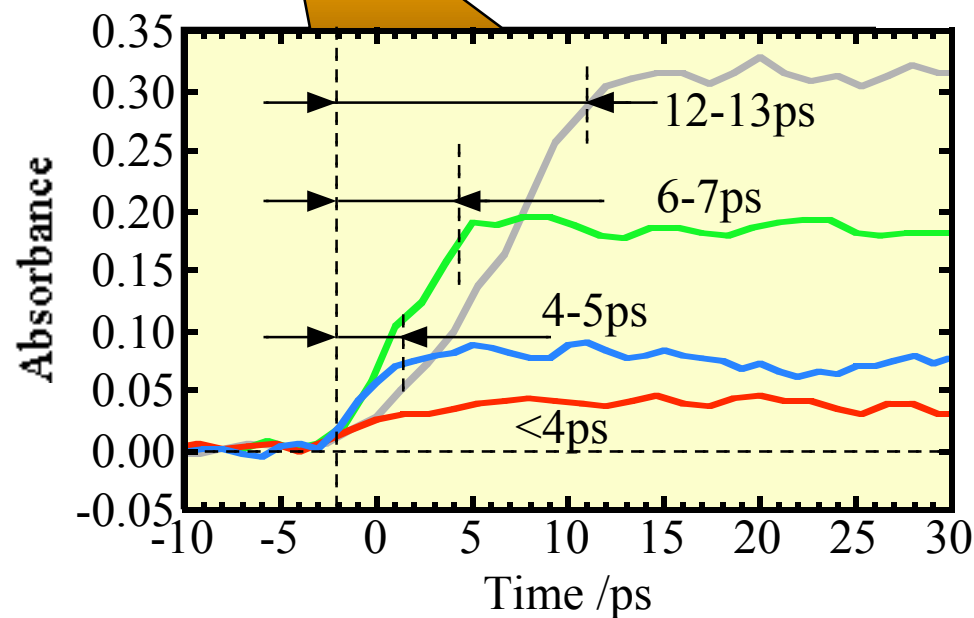
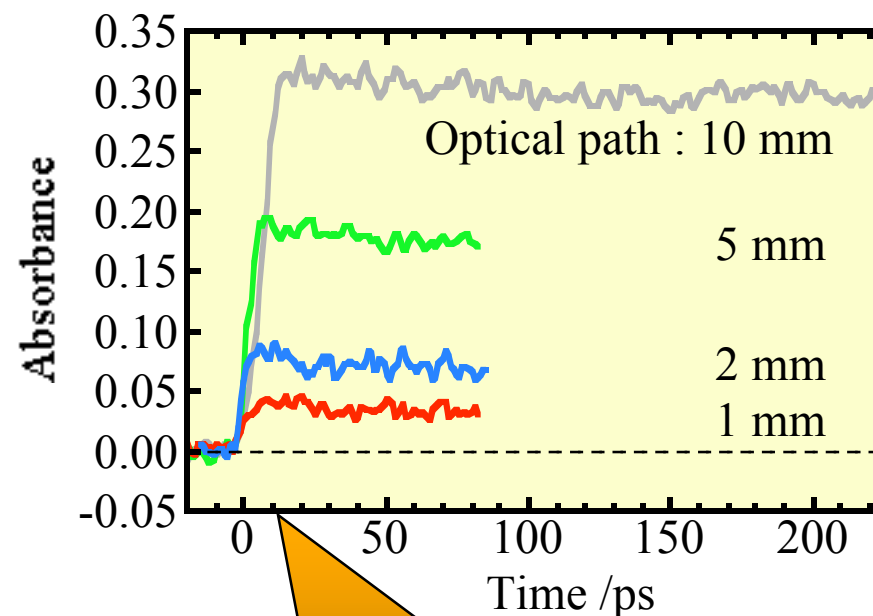
Time resol. /ps	12.2ps	7.2ps	5.2ps	3.2ps
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Time resolution: δ_{total}

$$\delta_{total} \approx \delta_{diff} + (\delta_E^2 + \delta_L^2 + \delta_{sync}^2)^{1/2}$$

Dominant factor : δ_{diff}

due to refractive index $n=1.33$



Comparison with conventional linac

	Conventional linac		Photocathode	
	Long-pulse mode	Single-pulse mode	In 2000	Current
Charge	8~20nC	0.5~0.6nC	0.8~1.0nC	1.7~2.0nC
Beam size	15x6mm	4mm	4mm	3mm
Dose/shot	30~50Gy	7~8Gy	13~15Gy	>40Gy
Pulse width	10-100ns	10ps	7ps	3ps

Dominant factors for time resolution

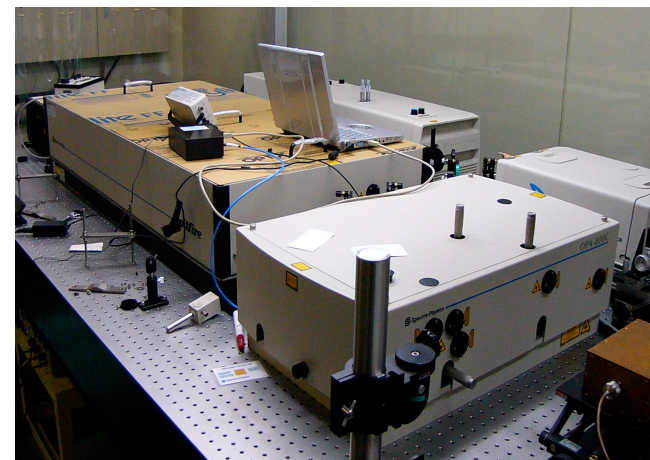
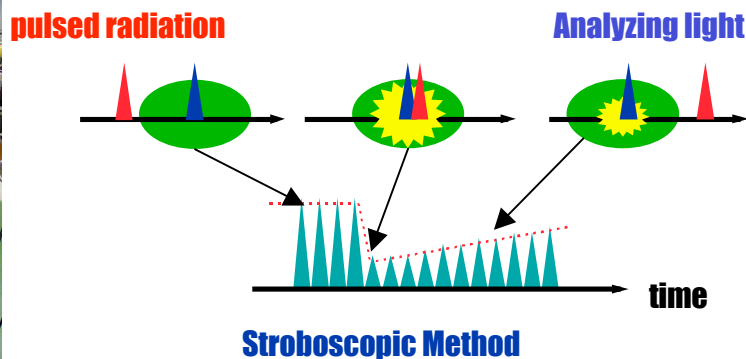
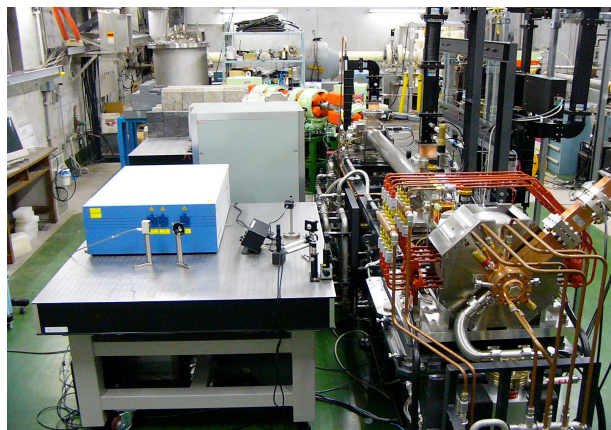
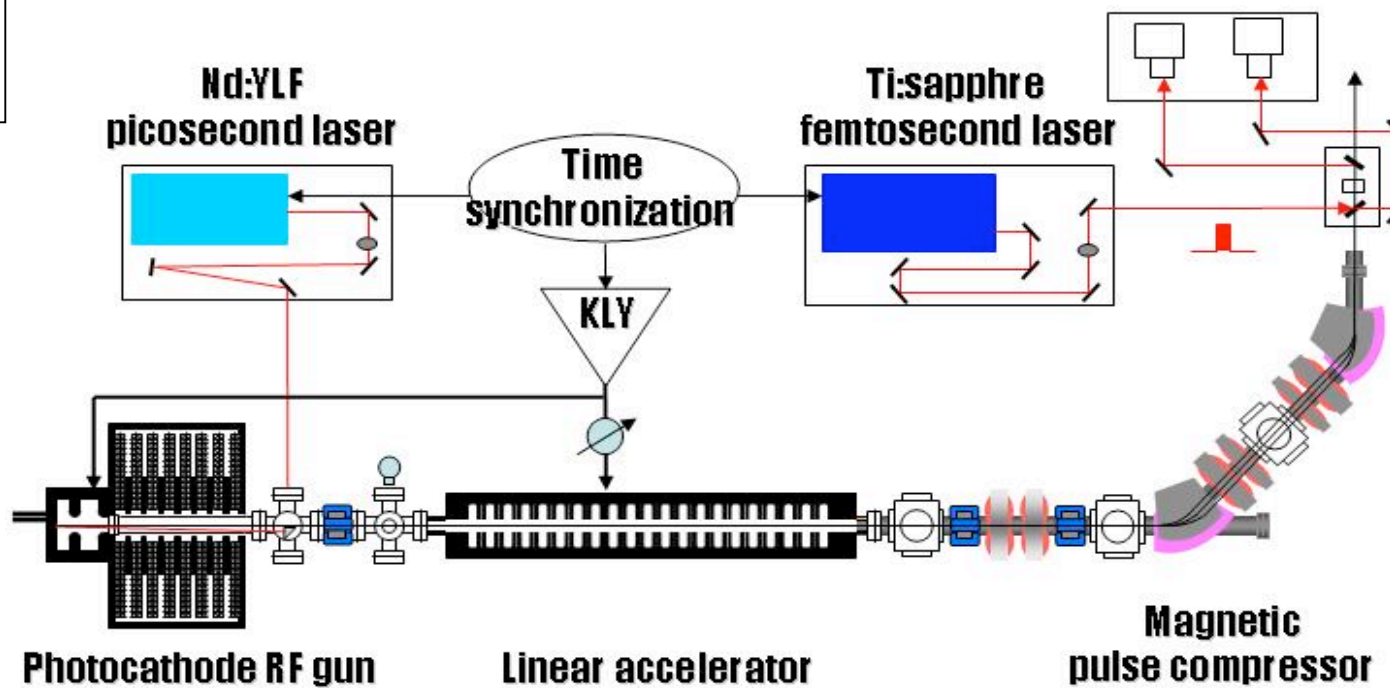
- (1) <3ps : pulse width of EB
- (2) 100fs : pulse width of laser
- (3) <1ps : synch. jitter

Not so bad

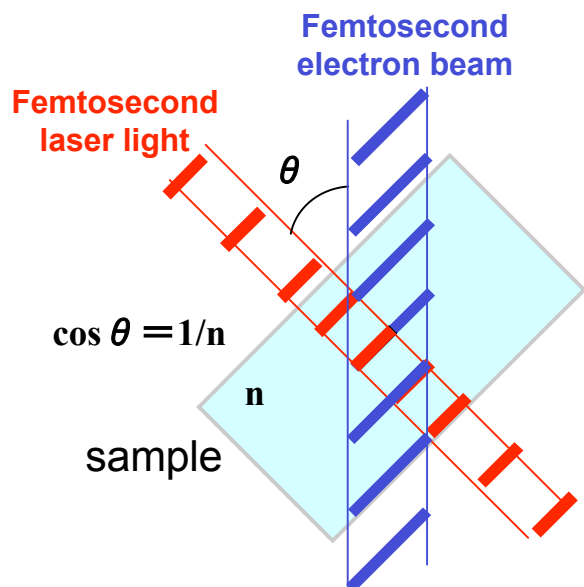
(4) **10ps /10mm** : Δt_{EB-L} in H_2O → Thinner cell, but OD ↓ → **High-Brightness EB**

Femtosecond electron beam and femtosecond pump-probe experiment in Osaka University

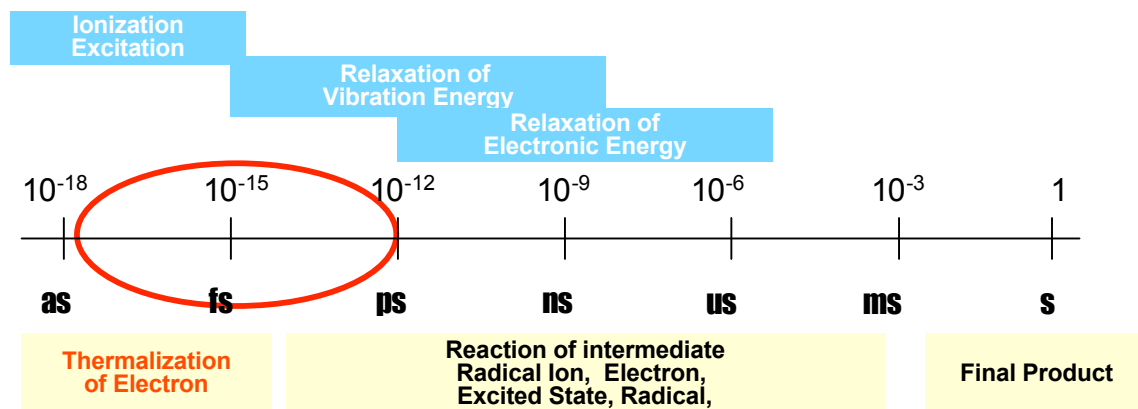
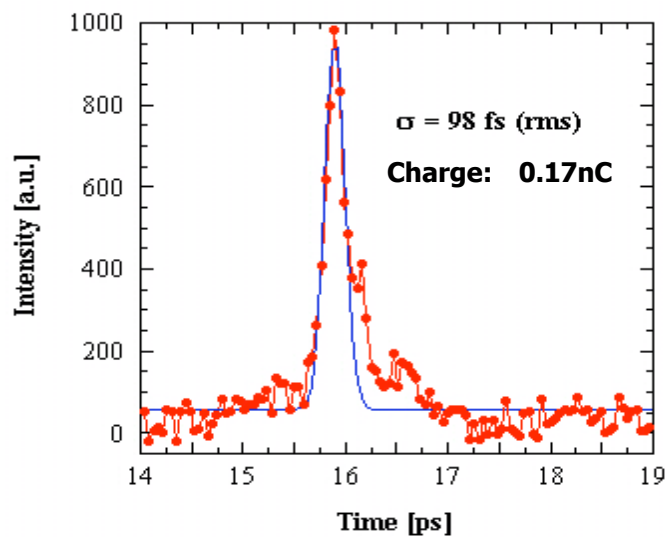
Osaka Univ.



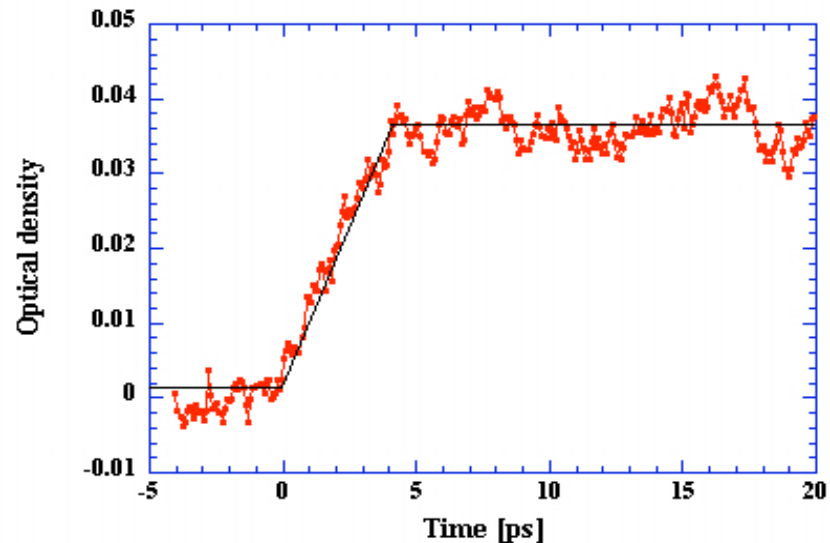
A new concept of “equivalent velocity spectroscopy” for studies of ultrafast electron-induced reactions



Temporal distribution of 98 fs electron bunch measured by the streak camera at 0.17 nC

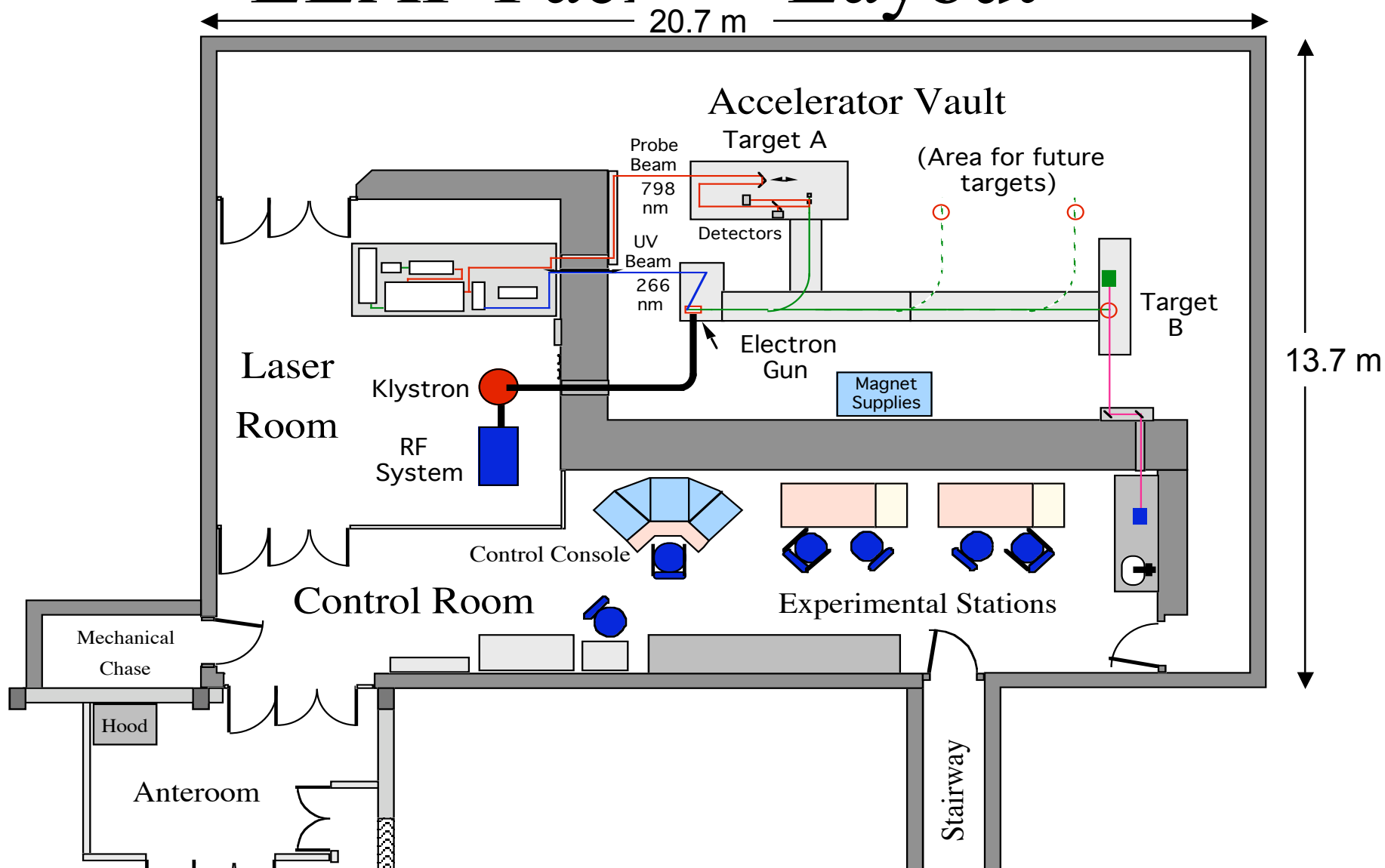


Transient absorption kinetics of hydrated electrons measured in water at wavelength of 800 nm



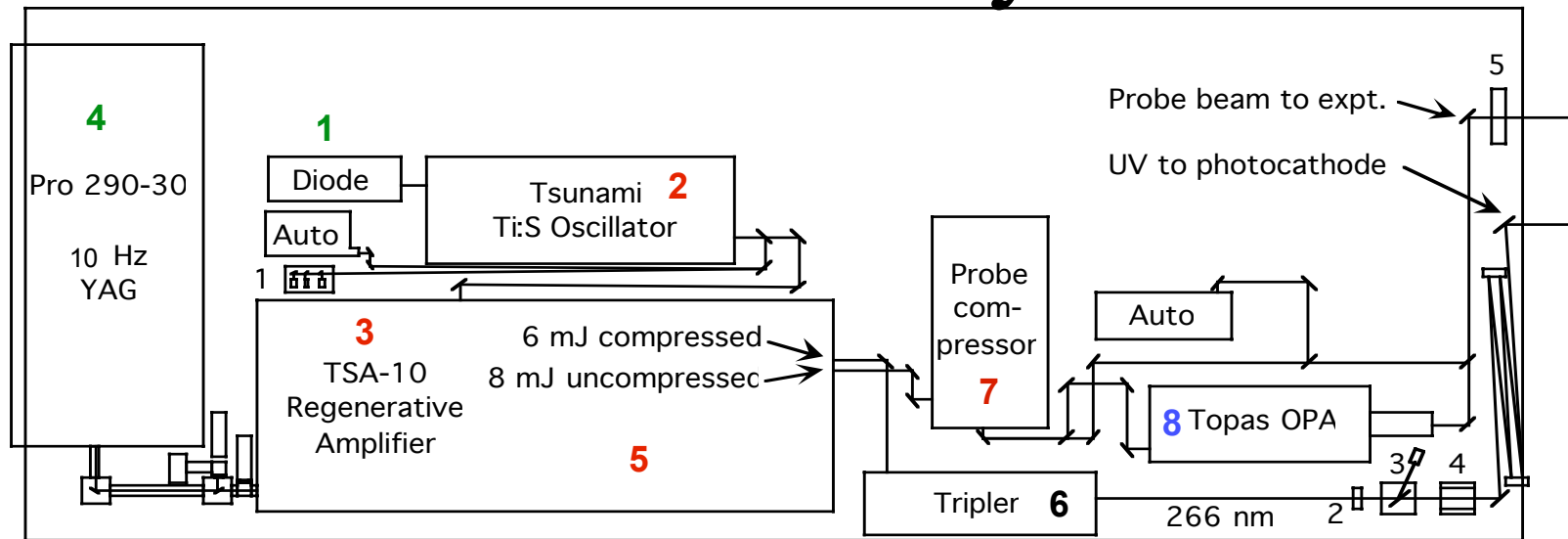
LEAF

LEAF Facility Layout



LEAF

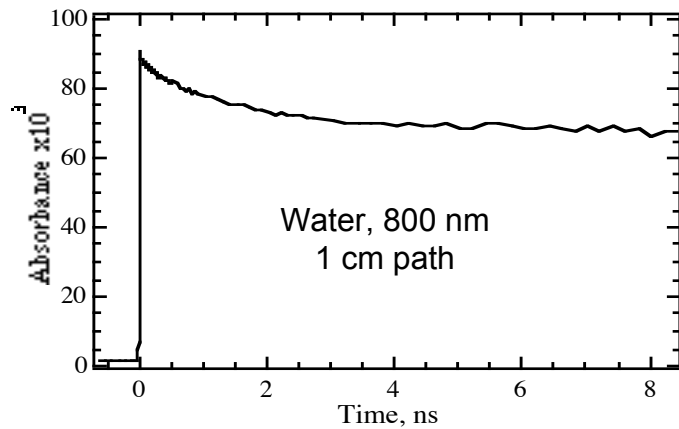
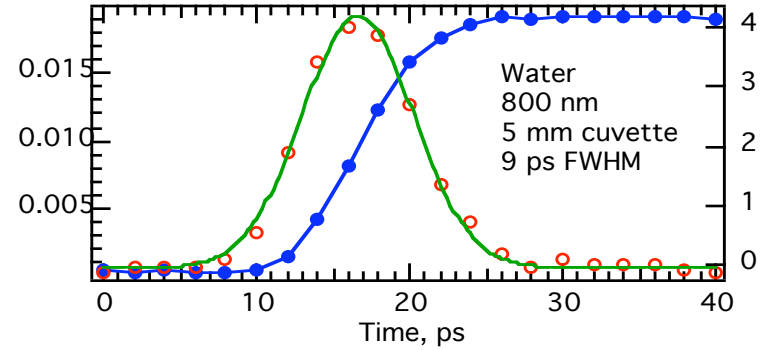
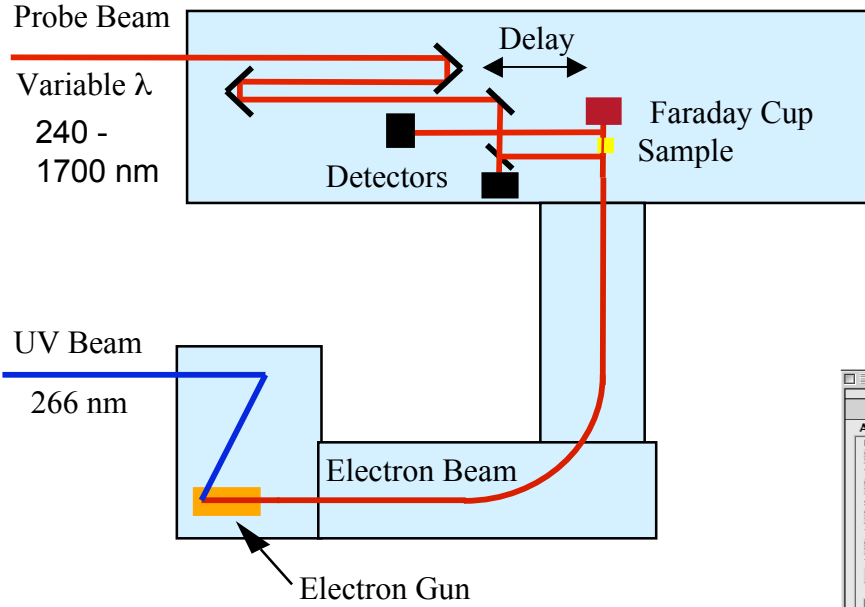
LEAF Laser System



- 1) Diode-pumped Nd:YVO₄ laser, 5 Watts, 532 nm, pumps picosecond Ti:Sapphire laser.
- 2) Ti:Sapphire oscillator produces ~50 fs pulses, ~ 7 nJ energy, 798 nm, at 81.60 MHz.
- 3) Pulse stretcher stretches oscillator pulse to > 200 ps, then injects the pulse into the Ti:Sapphire regenerative amplifier.
- 4) Simultaneously, the doubled, Q-switched Nd-YAG laser pumps the Ti:Sapphire regen.
- 5) Stretched ~200 ps pulse is amplified to ~12 mJ level. Half is compressed to 1-3 ps for THG
- 6) 1-3 ps pulse is frequency tripled to 266 nm (≤ 0.4 mJ) for excitation of Mg photocathode.
- 7) Half of regen output compressed to ~100 fs for use as probe or TOPAS OPA pump (8)

LEAF

Pulse-Probe Experiment



Acquisition parameters

- Number of Shots per Point: 20
- Time Interval per Point (ps): 200.0
- Starting Delay Time (ns): -0.030
- Total Time Window (ns): 9.800
- Time Window Used (ns): 9.800
- Number of Points: 50
- Iteration: 136
- Cumulative shots: 2720
- Scope Trigger Rep Rate (Hz): 10.0

I/O parameters

- Root Data File Name: CD3CN
- Data Folder Path: C:\Macintosh HD\Users\Jim\041108 NIR
- Next Data File is: CD3CN_16
- File header/comment: CD3CN, aerated, still, settings 041110-s1, phase 16.3, 8.75 MeV, 3.1 nC

Plots:

- Absorbance:** Shows a sharp peak at approximately 1.5 ns, decaying to a baseline of about 0.025 by 2 ns.
- Corrected Absorbance:** Shows a similar peak at 1.5 ns, but with a different baseline and decay characteristics.
- Faraday Current:** Shows a step-like increase at 1.5 ns, indicating the arrival of the electron beam.

LEAF Pulse-probe transient absorption spectroscopy

LEAF

Time resolution ≥ 7 ps

A factor of pulse width and sample depth.

Optical Parametric Amplifier (240 - 2600 nm)

New diodes extend range from 1000 to 1700 nm

Color separation needs work, far-field mode varies

Better Signal/Noise than before

Improved LabVIEW acquisition software

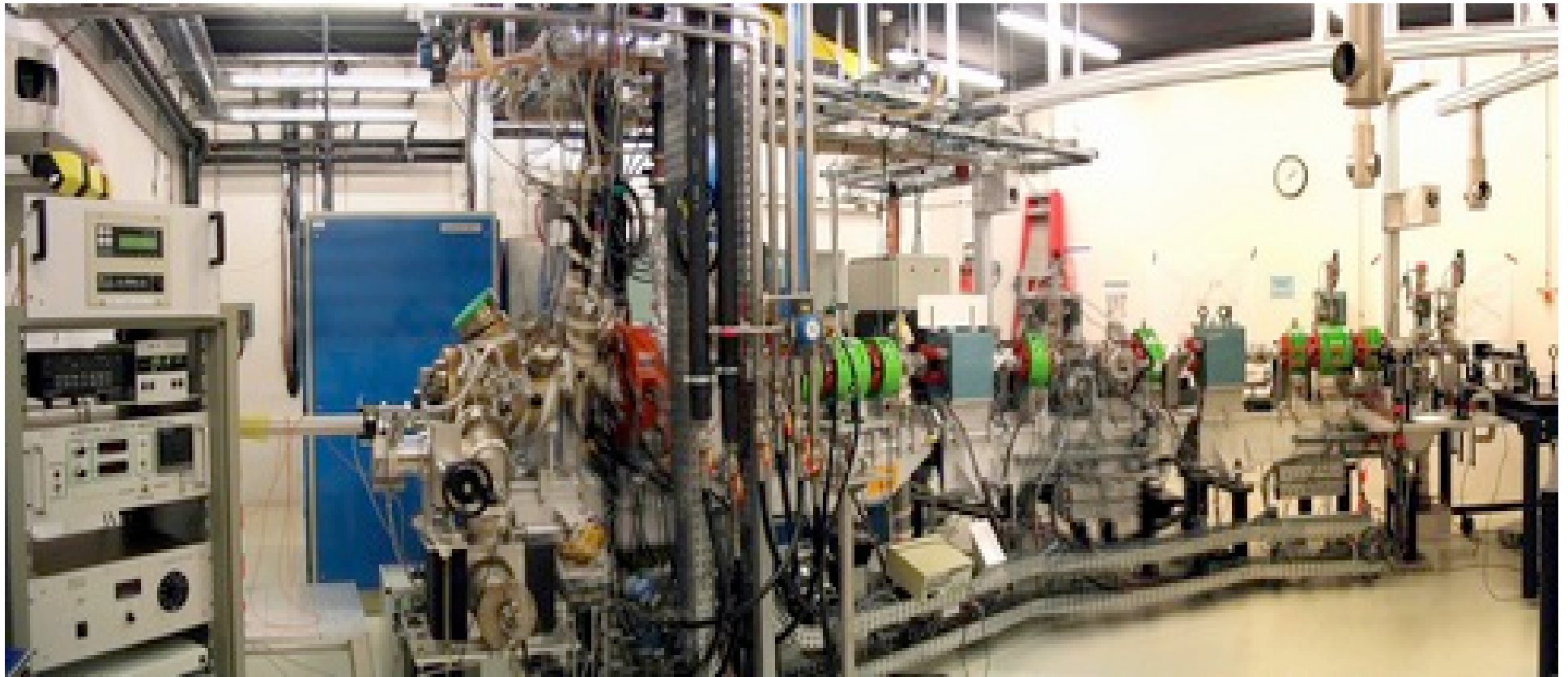
Interleaved collection, measurement selection criteria (dose, laser intensity, modulator and laser timing), using consistent cathode recovery time

More proficient laser and electron beam alignment

Igor-based analysis software

Flow system volumes reduced (~ 4 ml or ≥ 15 ml)

ELYSE, Orsay



ELYSE, Picosecond Pulse Radiolysis

ELYSE, Orsay

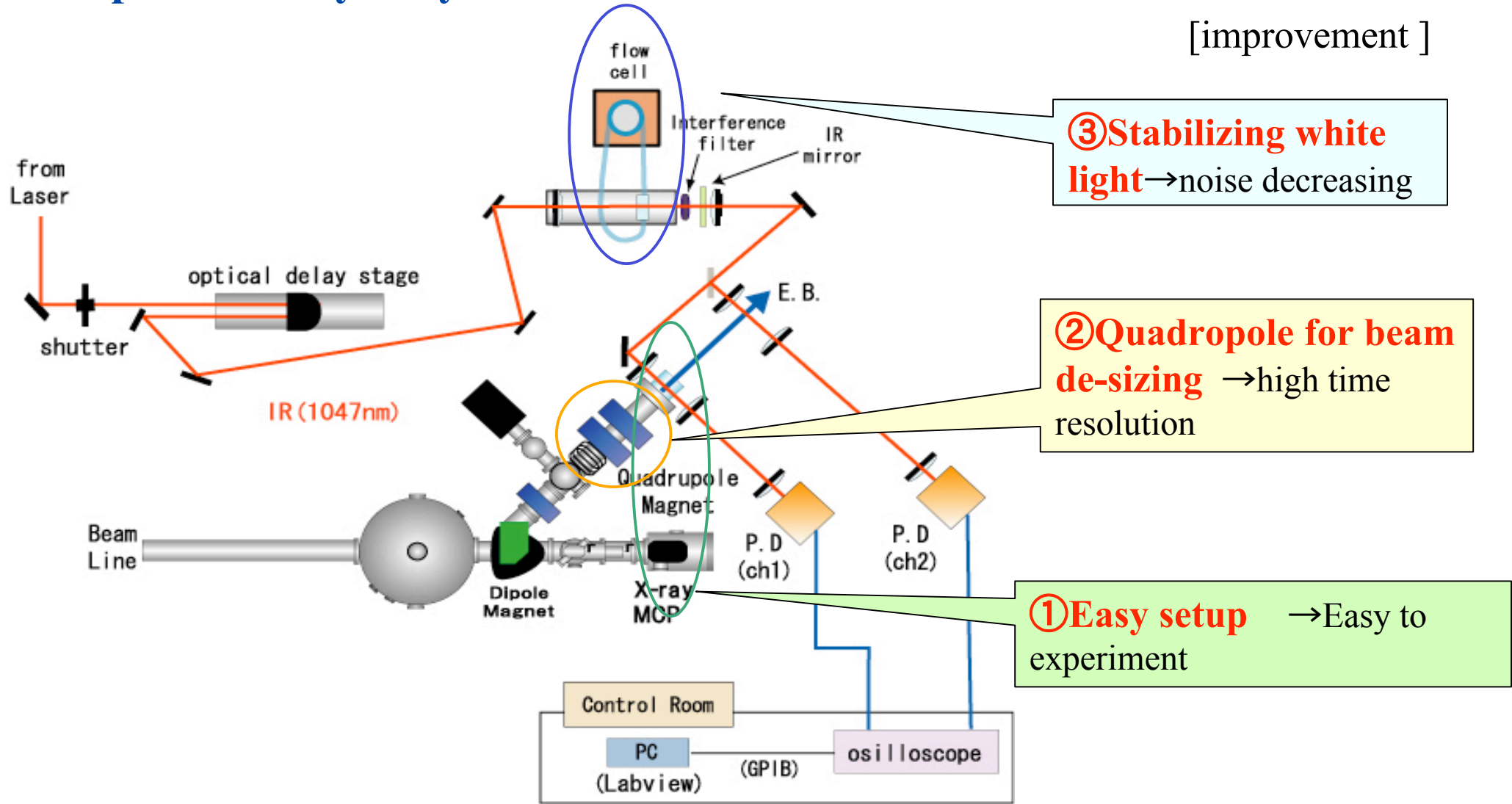
Photoinjector Accelerator

- Pulse length ≤ 7 ps
- ♣ Charge ≥ 1 nC
- ♣ Energy 4 to 9 MeV
- ♣ Repetition Rate ≤ 50 Hz
- ♣ Energie Dispersion $\leq 2,5$ %
- ♣ Spot Diameter ≤ 2 à 20 mm
- ♣ Pulse-Probe

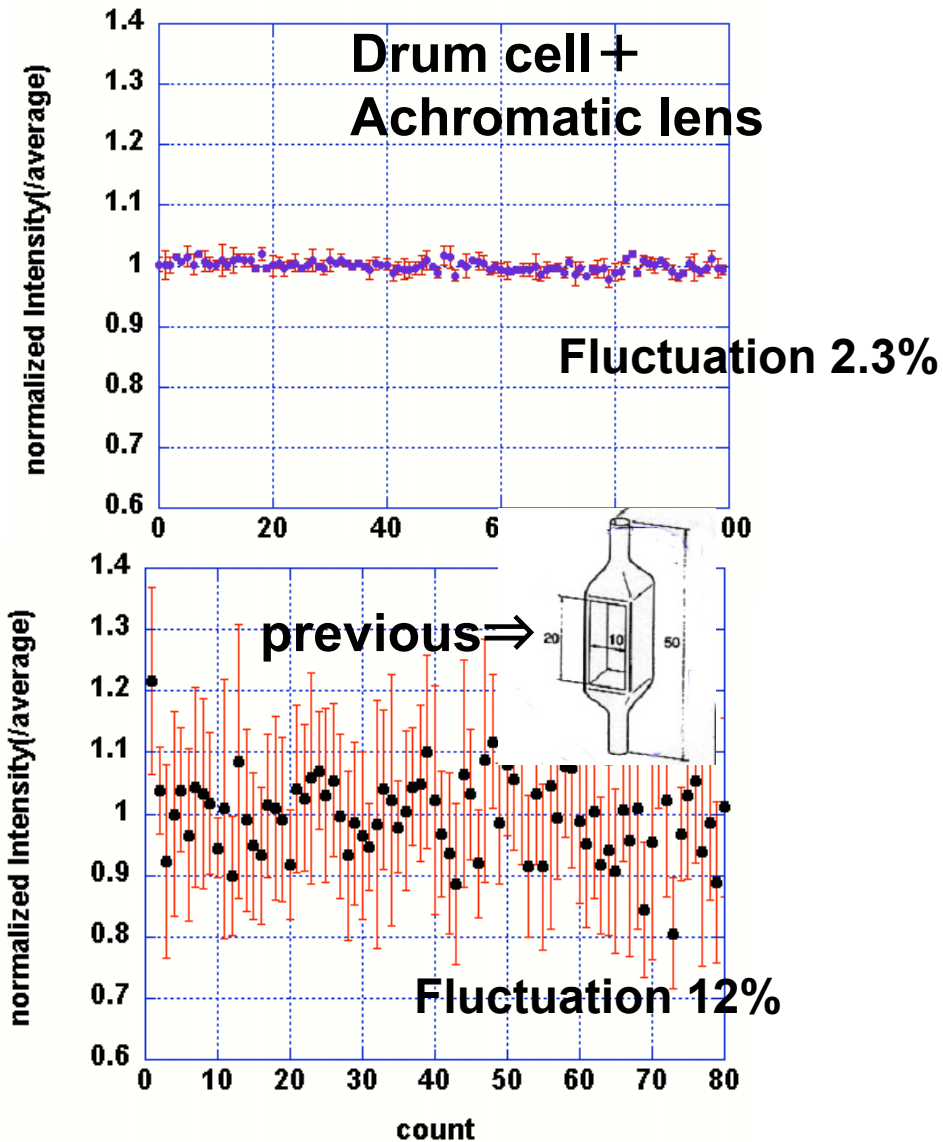
Accelerator build and installed (SERA)



New pulseradiolysis system



Stability of Probe light intensity



Experimental results (right water)

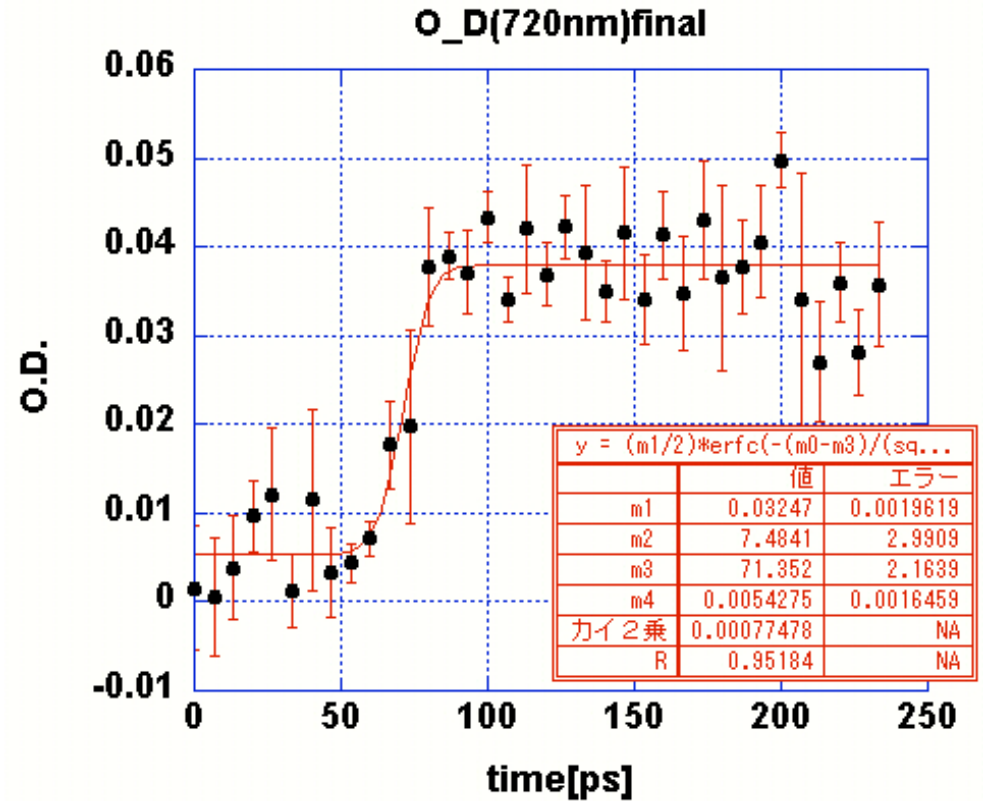


Fig.12 O.D.

previous (26[ps]) : Current 8[ps]

	Beam energy	Beam Current	Beam width	Beam size	Target path Length	Synchro-nization	Laser pulse width	Total time Resolution
U. Tokyo	4+18= 22MeV	2nC	1ps	3mm	1mm	<1ps(rms)	100fs(532nm-2600nm)OPA (400-1100nm) white light made by Ti:Sa	3ps(white light)
LEAF,BNL, USA	9MeV	2-8nC	≥ 7 ps		10mm(right water)	Pico-sec.	100fs(240-2600nm)OPA	>7ps(pulse-probe)
ELYSE, France	4 to 9 MeV	≥ 1 nC	≤ 7 ps	2-20mm				~ 7 ps?
Waseda Univ.	4MeV	0.4-0.6nC						8ps
Osaka Univ.	38MeV	>0.2nC	<1ps				100fs	~ 5 ps

Summary

Photocathode RF gun with fs laser(Tt:Sa) is suitable combination for the Application of Radiation Chemistry

In order to measure the phenomena at sub-pico or picosecond region, we need;

- high brightness beam with short pulse(<1ps)
- Thin target(~mm)
- Stable system
 - Timing (within 1ps)
 - Position
 - Beam Intensity (both laser and electron beam)

Special thanks to

LEAF, BNL

ELYSE, Orsay

Univ. Osaka

Waseda Univ.

Univ. Tokyo

Physicochemical stage

- Factor of the thermalization distance

$$\lambda_m = (n\sigma)^{-1}$$

$$\lambda = [-\lambda_m \ln (P \lambda_m)]/\alpha$$

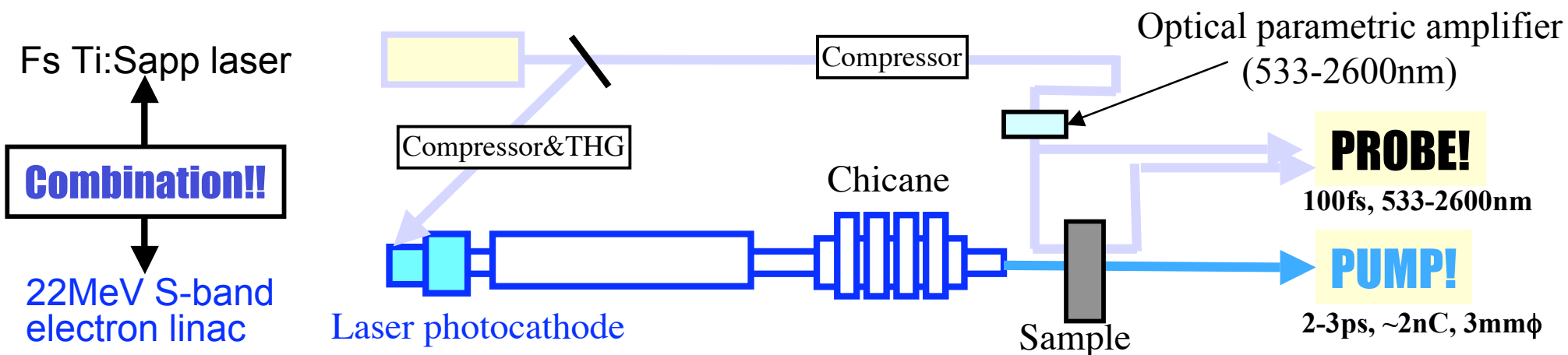
- Factor of the cross section of geminate ion recombination

$$\sigma_{rec} \sim 3 \times 10^{-20} E^{-1.86} \times \beta$$

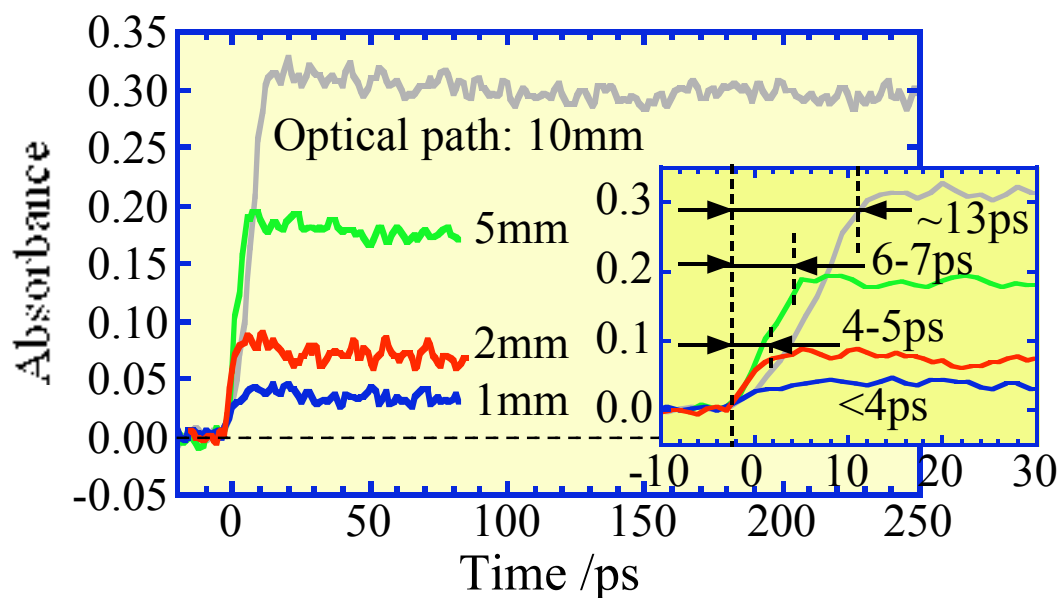
- Branching ratios in physicochemical stage stems from ionization and excitation

	Probability		Probability	Reaction No.
Decay of the directly excited water molecules				
H ₂ O*(A ¹ B ₁) → H ₂ O	P1			(1)
→ H + OH	P2			
H ₂ O*(B ¹ A ₁) → H ₂ O ⁺ + e ⁻	P6			(4)
→ H ₂ O	(1-P6)P1			(1)
→ dissociation	(1-P6)P2	→ H + OH	P3	(2-1)
		→ 2H + O(3P)	P4	(2-2)
		→ H ₂ + O(¹ D)	P5	(2-3)
Decay of the excited states results from the recombination				
H ₂ O* → H ₂ O	P1			(1)
→ dissociation	P2	→ H + OH	P7	(3-1)
		→ 2H + O(3P)	P8	(3-2)
		→ H ₂ + O(¹ D)	P9	(3-3)

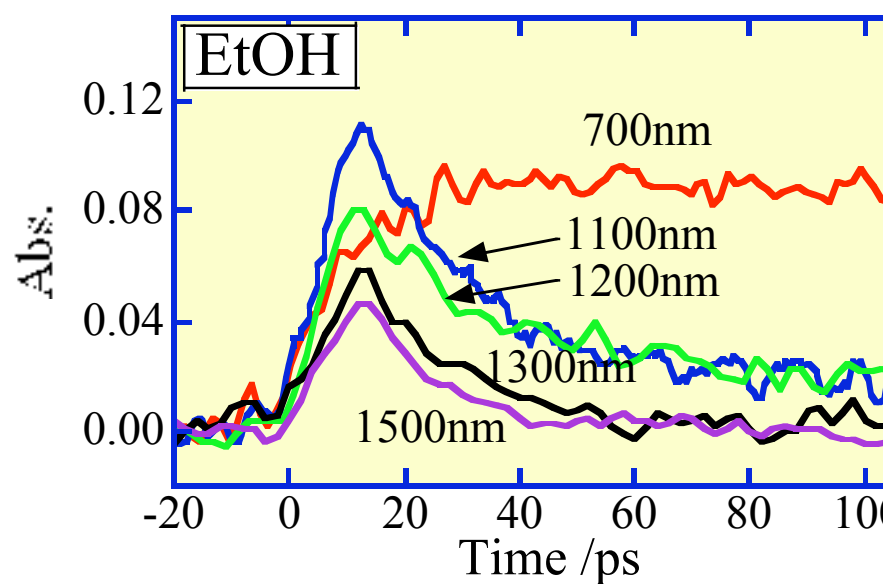
Ultra-fast pump-and-probe pulse radiolysis study : radiation induced fast processes



Time behaviors of hydrated electrons in water:
Solvation time < time resolution < 10ps

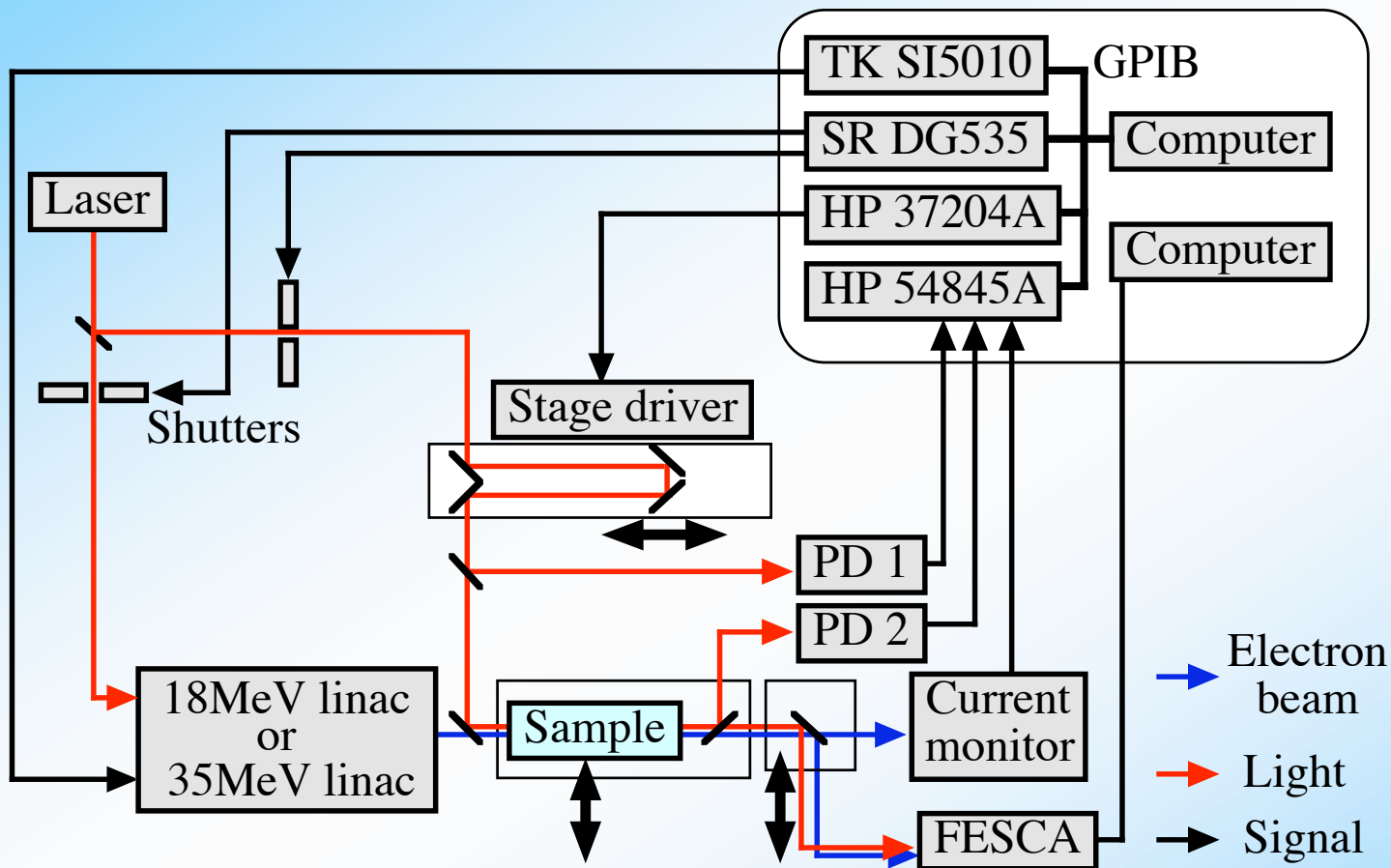


Time behaviors of solvated electrons in ethanol:
Observation of solvation process ($e^-_{pre} \rightarrow e^-_{sol}$)



Sub-ps Pulse Radiolysis - Measurement System

Beam-Material Interactions, UTNs



Data acquisition

- Measurement of laser intensity and charge

- **B** : Both beam and light $\rightarrow I_M(B)$ and $I_R(B)$
- **L** : Light only $\rightarrow I_M(L)$ and $I_R(L)$
- **P** : Beam only $\rightarrow I_M(P)$ and $I_R(P)$
- **N** : Neither beam nor light $\rightarrow I_M(N)$ and $I_R(N)$
- **Charge** $\rightarrow C$

(I_M : Main light, I_R : Reference light)

- Calculation of precise absorbance

$$Absorbance = \log_{10} \frac{I_0}{I} = \frac{C_{ave}}{C} \cdot \log_{10} \left[\frac{I_M(L) - I_M(N)}{I_R(L) - I_R(N)} \cdot \frac{I_R(B) - I_R(P)}{I_M(B) - I_M(P)} \right]$$

(C_{ave} : Average of charges)