

Spring 1997 #5

p137

A binary alloy is composed of N_A atoms of type A and N_B atoms of type B. Over the temperature range of interest, the A type atoms can only be found in the ground state and in an excited state of energy E_0 . Similarly, the B-type atoms can only occupy the ground state and an excited state having energy $2E_0$. The entire mixture is in thermal equilibrium at temperature T .

a) Calculate the Helmholtz free energy F of this alloy.

→ Since the ground state energy is unspecified, assume it is zero for both A and B type atoms.

$$Z = (1 + e^{-\beta E_0})^{N_A} (1 + e^{-2\beta E_0})^{N_B}$$

$$F = -kT \ln Z = -kT N_A \ln(1 + e^{-\beta E_0}) - kT N_B \ln(1 + e^{-2\beta E_0})$$

b) Deduce the heat capacity of this system at temperature T .

find energy: E

$$E = -\frac{1}{\beta} \ln Z = -\frac{1}{\beta} [N_A \ln(1 + e^{-\beta E_0}) + N_B \ln(1 + e^{-2\beta E_0})]$$

$$\bar{E} = \frac{-N_A (-E_0) e^{-\beta E_0}}{(1 + e^{-\beta E_0})} - \frac{N_B (-2E_0) e^{-2\beta E_0}}{(1 + e^{-2\beta E_0})}$$

$$C = \left(\frac{\partial E}{\partial T} \right) = \left(\frac{\partial E}{\partial \beta} \right) \left(\frac{\partial \beta}{\partial T} \right)$$

$$C = \left[\frac{N_A E_0 e^{-\beta E_0} (-1) (-E_0) e^{-\beta E_0}}{(1 + e^{-\beta E_0})^2} + \frac{N_B E_0 (-E_0) e^{-\beta E_0}}{(1 + e^{-\beta E_0})} + \frac{2N_B E_0 e^{-2\beta E_0} (-1) (-2E_0) e^{-2\beta E_0}}{(1 + e^{-2\beta E_0})^2} + \frac{2E_0 (-2E_0) e^{-2\beta E_0}}{(1 + e^{-2\beta E_0})} \right] \left(-\frac{1}{kT^2} \right)$$